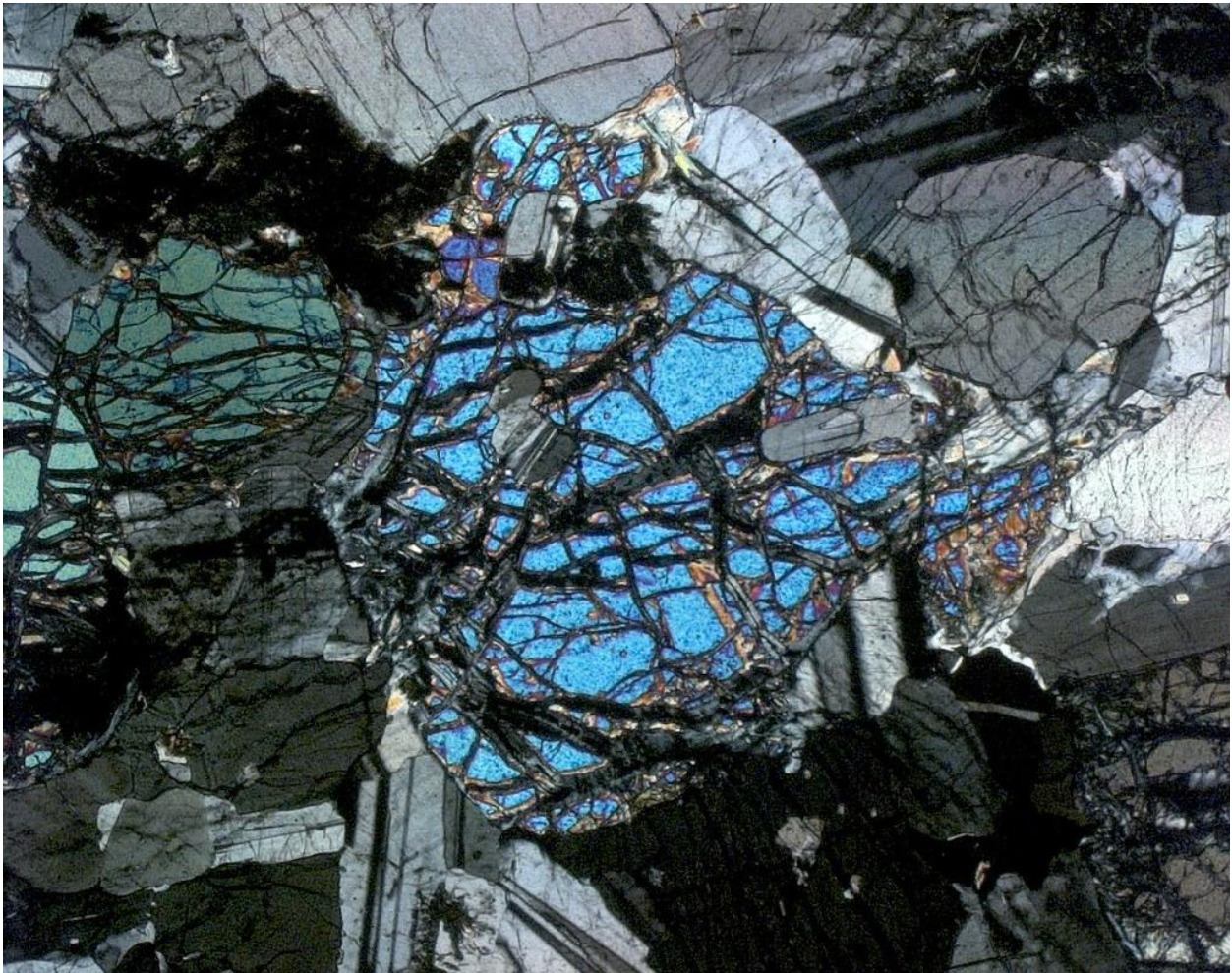


May 10, 2021

## Olivine



**Figure 1 - Olivine in Thin Section**

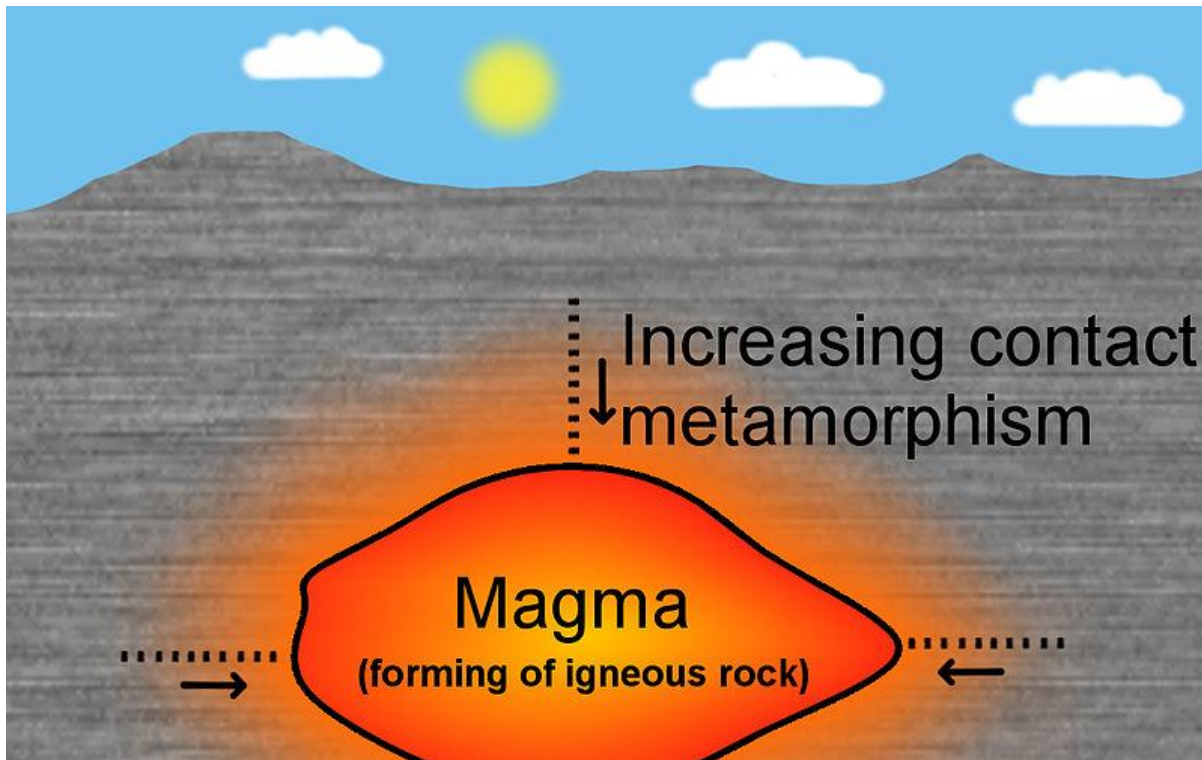
Credit: [only point five](#), "Olivine" by [only point five](#) is licensed under [CC BY-NC 2.0](#)

Olivine is an important rock forming mineral. It is also interesting in its own right. The [Olivine Group](#) includes the [Fayalite-Forsterite Series](#), [Monticellite](#), [Kirschsteinite](#), [Tephroite](#), and [Knebelite](#). The most common examples of olivine are found in the Fayalite-Forsterite Series, a solid solution series of iron-magnesium silicates. Replacing the iron and magnesium cations in the atomic matrix with calcium or manganese will make the other olivine minerals.

The distinguishing characteristics of olivine is its olive green colour, its [orthorhombic](#) crystal shape and moderate to imperfect cleavage.

Olivine minerals are typically found in [dunite](#), [peridotite](#), [gabbro](#), [dolerite](#), [basalt](#) and [syenite](#), but can also be found in sediments that have been [thermally metamorphosed](#) through contact with

an igneous intrusion. It is when rocks have been thermally metamorphosed that the iron and magnesium in Fayalite-Forsterite olivine can be replaced with calcium or manganese. Figure 2, below, graphically shows how thermal metamorphism works.



**Figure 2 - Thermal Metamorphism**

Credit: [Jasmin Ros](#), [Creative Commons Attribution-Share Alike 3.0 Unported](#)

Much of the [Earth's mantle](#) is made up of olivine and [olivine is also found in some meteorites](#). Researchers [Michael J. Russell](#) and [Adrian Ponce](#) listed olivine as one of the "[Six 'Must-Have' Minerals for Life's Emergence](#)".

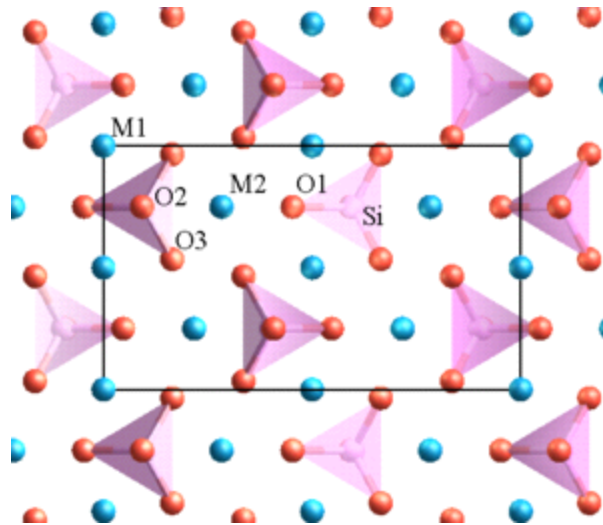
### Chemistry and Structure

The general chemical formula for olivine is



Where **A** is a cation such as iron ( $\text{Fe}^{2+}$ ), magnesium ( $\text{Mg}^{2+}$ ), calcium (Ca) or Manganese (Mn) and  $\text{SiO}_4^{4-}$  is the [silicate anion](#).

The general description of the structure of olivine is [orthosilicate](#). The silicate anion forms a tetrahedron that is not directly linked to other silicate tetrahedrons. Rather, the tetrahedrons are linked by the cations, as in Figure 3.



**Figure 3 - Atomic Structure of Olivine**

Credit: [andreww](#) , [Public Domain](#)

## Olivine Minerals

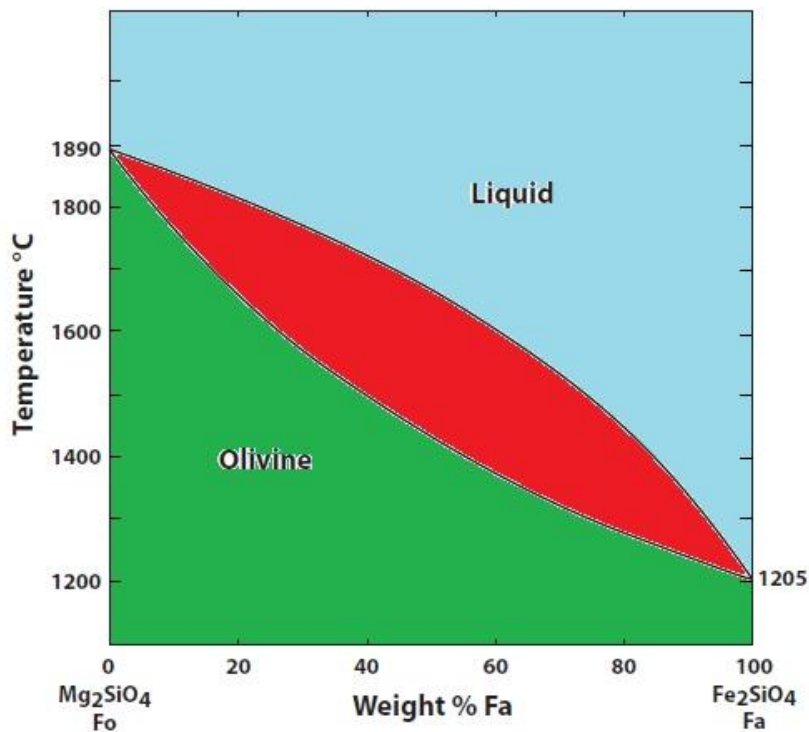
### [Fayalite-Forsterite Series](#)



**Figure 4 - Olivine**

Credit: [Aram Dulyan](#), [Public Domain](#)

The classic olivine minerals are those that are part of the Fayalite-Forsterite Series. Fayalite,  $\text{Fe}_2\text{SiO}_4$ , and forsterite,  $\text{Mg}_2\text{SiO}_4$ , form a solid solution series. Figure 5, below, shows a phase diagram for the Fayalite-Forsterite Series.



**Figure 5 - Forsterite-Fayalite Series**  
 Modified from [Binary Phase Diagrams](#)

Pure examples of forsterite are rare in nature, almost always there is a mixture of iron and magnesium in the crystal matrix. This is usually expressed as the forsterite percentage: the abbreviation Fo<sub>75</sub> means that the olivine is 75% forsterite. Olivine that leans toward the forsterite composition, Fo<sub>92</sub> and Fo<sub>88</sub>, is found in dunite and peridotite. Gabbro, dolerite, basalt, and [trachyte](#) often contain olivine in the Fo<sub>85</sub> to Fo<sub>40</sub> range.

Iron rich olivine and relatively pure fayalite are found in dolerite, [ferrogabbro](#), [granophyre](#) and syenite rocks. Fayalite also occurs in iron rich sediments that have been thermally metamorphosed.

Fayalite is named after the Island of [Faial](#) in the Azores. Forsterite is named after [Johann R. Forster](#), a German scientist and explorer.

### [Monticellite](#)



**Figure 6 - Monticellite Crystals on Brown [Vesuvianite](#)**

Credit: [Robert M. Lavinsky](#), [iRocks.com](#) – CC-BY-SA-3.0

Named after the Italian geologist [Teodoro Monticelli](#), Monticellite,  $\text{CaMgSiO}_4$ , is a gray to colourless mineral. It is most commonly found in thermally metamorphosed rocks at the contacts between volcanic rocks and [siliceous dolomitic limestone](#).

### [Kirschsteinite](#)



**Figure 7 - Kirschsteinite**

Credit: [Leon Hupperichs](#), [Creative Commons Attribution-Share Alike 3.0 Unported](#)

Kirschsteinite,  $\text{CaFe}^{2+}\text{SiO}_4$ , is a greenish grey olivine mineral where calcium has been incorporated into the crystal matrix. It is typically found in calcareous [skarn](#), another thermally metamorphosed rock formed in the contact between intrusive volcanic rock and limestone sedimentary rock. Kirschsteinite was named after the German geologist Dr. Egon Baron von Kirschstein, best known for a disastrous 1908 expedition to [Mount Karisimbi](#) on the border between Rwanda and the Democratic Republic of the Congo.

### [Tephroite](#) and [Knebelite](#)



**Figure 8 - Tephroite**

Credit: [Robert M. Lavinsky](#), [iRocks.com](#) – CC-BY-SA-3.0

Varying in colour from grey, to olive-green, to flesh red or reddish-brown, to dark brown and black, Tephroite,  $\text{Mn}_2\text{SiO}_4$ , and closely related Knebelite,  $(\text{Mn Fe})_2\text{SiO}_4$ , are formed where manganese is incorporated into fayalite. These two minerals are also found skarns formed in the contact between intrusive volcanic rock and [iron-manganese ores](#).

Tephroite was named by [Johann Friedrich August Breithaupt](#) for its colour, from the Greek τειφροζ "tephros" = ash-coloured. Knebelite was named after a German poet and translator, [Karl Ludwig von Knebel](#).

### **Olivine as Jewelry**



**Figure 9 - Swarovski Tennis Bracelet in Olivine**

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Olivine is best known as the jewel [Peridot](#). Peridot serves as a birthstone for the month of August. The most valuable Peridot crystals are dark green or lime green and are relatively clear. Gem quality olivine tends to be on the Forsterite end of the Forsterite-Fayalite Series.

### **Standard Caveat**

The purpose of my weblog postings is to spark people's curiosity in geology. Don't entirely believe me until you've done your own research and checked the evidence. If I have sparked your curiosity in the subject of this posting, follow up with some of the links provided here. If you want to, go out into the field and examine some rocks on your own with the help of a good field guide. Follow the evidence and make up your own mind.

**In science, the only authority is the evidence.**